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Definition Of Supersaturated Solution

Definition Of Supersaturated Solution

A supersaturated solution is a solution that contains more than the average solvent that can be dissolved at a given temperature. The recrystallization of the excess dissolved solvent in a supersaturated solution can be started by inserting

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a tiny solute crystal,
called a seed crystal.

Supersaturated Solution - Definition, Examples ...

A supersaturated solution is one that is forced to hold more atoms in water or another solute than it normally would. Design and create your world with stem. It can be caused by reduction of the emulsion globules linear size that

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contains ammonium
nitrate supersaturated
solution N [H.sub.4]N
[O.sub.3].

Supersaturated solution - definition of supersaturated ...

Supersaturation is a solution that contains more of the dissolved material than could be dissolved by the solvent under normal circumstances. It can also refer to a vapor of a compound that has a

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higher (partial)
pressure than the
vapor pressure of that
compound.

Supersaturation is used
in cloud chambers.

Supersaturation - Wikipedia

supersaturated
solution. 1) A solution
that contains a higher
than saturation
concentration of
solute; slight
disturbance or seeding
causes crystallization

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Definition of Supersaturated Solution - Chemistry Dictionary

supersaturated
solution an unstable
solution containing
more of the solute than
it can permanently
hold.

Supersaturated solution | definition of supersaturated ...

Luckily for you,
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chemists know how to make a supersaturated solution, a solution that holds more solute than it normally could in its saturated form. Supersaturated solutions are very unstable, and...

Supersaturated Solution: Definition & Example - Video ...

Supersaturated definition, to increase the concentration of (a solution) beyond

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Solution
saturation; saturate
abnormally. See more.

Supersaturated | Definition of Supersaturated at ...

In this case, a supersaturated solution of sodium acetate is poured over a crystal of sodium acetate. These crystals provide the lattice structure "seed" which causes the sodium acetate ions in solution to crystallize out. The

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salt begins to crystallize, forming a large sodium acetate structure from the precipitation of the ions out of solution.

10.16: Saturated and Supersaturated Solutions - Chemistry ...

It's produced from a supersaturated solution of water and sodium acetate crystals. When you introduce a single

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sodium acetate crystal into such a solution, it's like a trigger that initiates a sudden reaction whereby all the excess sodium acetate in solution quickly crystallizes.

How to Make a Supersaturated Solution | Sciencing

tr.v. su·per·sat·u·rat·ed,
su·per·sat·u·rat·ing,
su·per·sat·u·rates 1. To
cause (a chemical
solution) to contain

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**Supersaturated
Solution**
more dissolved solute
than is normally
possible under given
conditions of
temperature and
pressure. 2.

Supersaturated - definition of supersaturated by The Free ...

Definition of
supersaturated :
containing an amount
of a substance greater
than that required for
saturation as a result

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Solution

of having been cooled
from a higher
temperature to a
temperature below
that at which
saturation occurs a
supersaturated
solution air
supersaturated with
water vapor Examples
of supersaturated in a
Sentence

**Supersaturated |
Definition of
Supersaturated by
Merriam**

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**Definition Of
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Solution**

A supersaturated solution, on the other hand, is when the excess of solute is dissolved in the solvent as a result of changes in temperature, pressure or other conditions. At room temperature, a saturated solution keeps the maximum possible amount of solute, and the rest becomes excess.

Unsaturated vs
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Definition Of Saturated vs Supersaturated solutions ...

The definition of a supersaturated solution is one which contains more dissolved solute than could ordinarily dissolve into the solvent. A minor disturbance of the solution or introduction of a "seed" or tiny crystal of solute will force crystallization of excess solute.

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Definition Of
**Saturated Solution
Definition and
Examples**

adjective 1. (of a solution) containing more solute than a saturated solution and therefore not in equilibrium 2.

**Supersaturated
definition and
meaning | Collins
English ...**

Supersaturate |
Definition of

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Supersaturate by
Merriam-Webster.

Supersaturate
definition is - to add to
(a solution) beyond
saturation. to add to (a
solution) beyond
saturation.... See the
full definition.
SINCE1828.

**Supersaturate |
Definition of
Supersaturate by
Merriam-Webster**
supersaturated Current
theory suggests that

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the first step in crystal growth is the formation of nuclei of a critical size in a supersaturated solution. From Cambridge English Corpus Although the discrepancy is small, it is particularly significant that in the experiments the melt became supersaturated, a possibility not allowed for in the theory.

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SUPERSATURATED | meaning in the Cambridge English Dictionary

Supersaturated solutions are formed if you add additional solute to the saturated solution. In other words, it is the condition in a saturated solution, when you add some additional quantity of solute into the solution.

Difference Between

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Definition Of Supersaturated Solution ...

In a supersaturated solution, there is more dissolved solute than in a saturated solution. The solute can easily fall out of solution by crystallization or precipitation. Special conditions may be needed to supersaturate a solution. It helps to heat a solution to increase solubility so

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more solute can be
added.

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